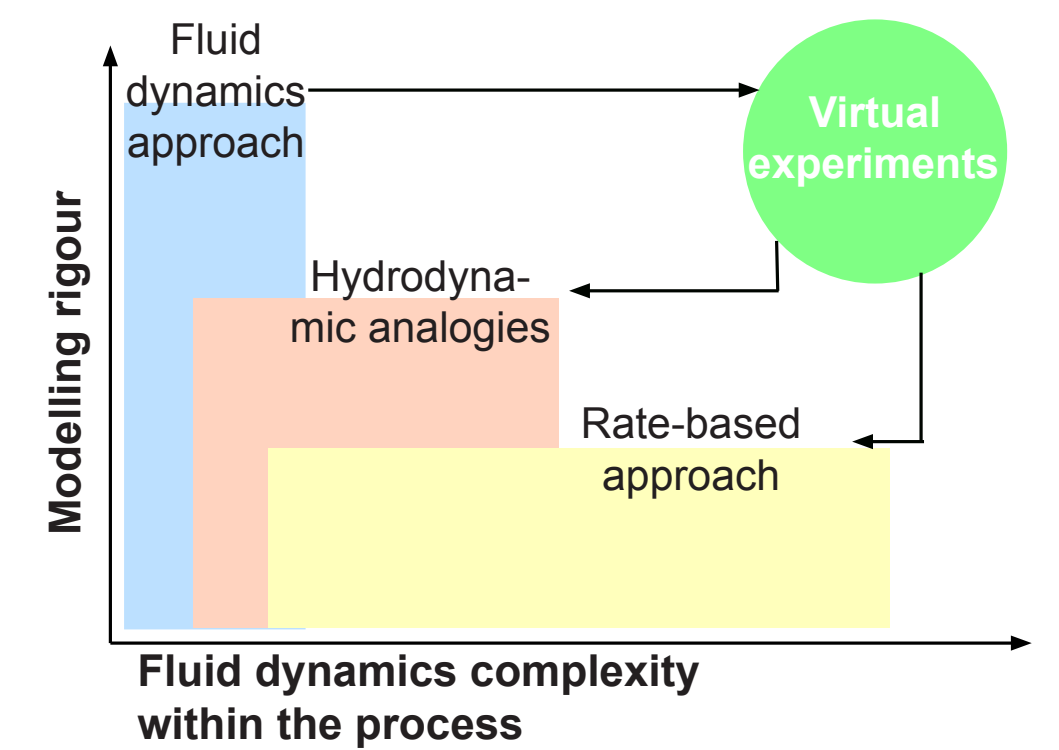


Complementary Modelling of Fluid Separation Processes

Principle

Despite that different fluid separations have much in common, a unified approach to their modelling is missing. This is mainly due to the diversity of operating, scale and boundary conditions as well as due to the significant modelling difficulties. A unified modelling approach is particularly difficult to formulate when a direct account of the process rates (transport and reaction kinetics) is essential, which is common in design and optimisation tasks. Therefore, we suggest an alternative way and present a novel modelling methodology which comprises different specific kinetics-based approaches and combine them in a complementary way. Such complementary modelling is a promising means to support the development of sustainable chemical engineering processes.



Outline and Main Results

Fluid-dynamics approach

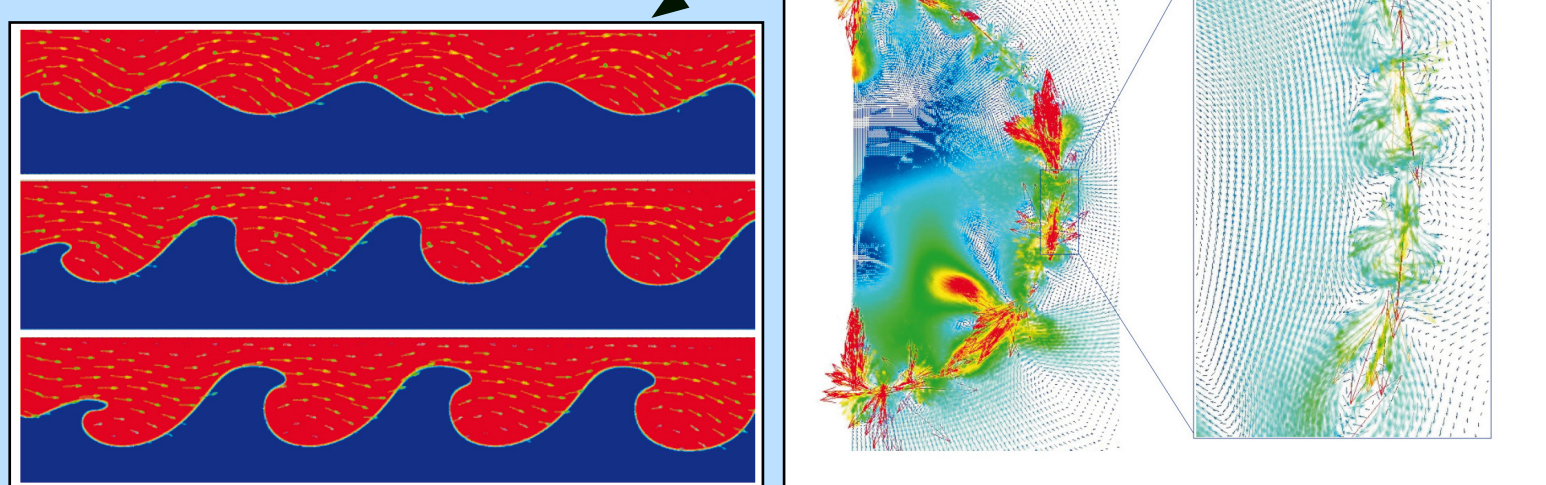
Idea: Process description by partial differential equations of fluid dynamics

Characteristics:

- Provides full information about the process (velocity, temperature and concentration field) in a purely theoretical way
- Difficult to apply for complex multiphase flow patterns

Results obtained:

- Film flows (cocurrent and countercurrent, laminar and turbulent, binary and multicomponent, one-, two- and three-phase, non-reactive)
- Liquid droplet and layer phenomena
- Monoliths
- Microseparations



T. Atmakidis & E. Y. Kenig, 2007, Escape17, Bucharest, Romania

Virtual experiments

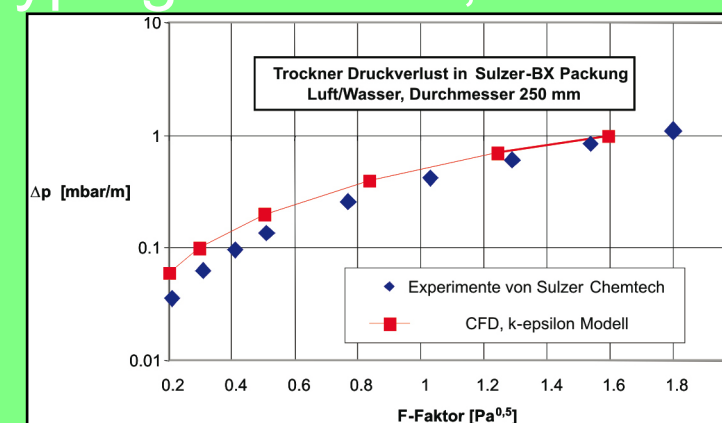
Idea: Replacement of empirical parameter estimation by model-based simulations for different basic types of column internals

Characteristics:

- Extracts information from the investigation of small periodic representative elements of real internals (usually via CFD)
- Provides a link between FDA and RBA/HA
- Potentially, a way to virtual prototyping internals, however still in the early stage

Results obtained:

- Pressure drop in fixed beds and structured packings
- Residence time distribution in catalytic packings
- Gas-liquid mass transfer coefficients in catalytic packings
- Liquid-solid mass transfer coefficients in fixed catalyst beds



Y. Egorov, F. Meiner, M. Klotter & E. Y. Kenig, 2005, Chem. Eng. Process., 44, 631-644

Hydrodynamic analogy approach

Idea: Development of analogies between real complex dynamics and geometrically simpler flow patterns based on experimental observations

Characteristics:

- Allows an extension of the application of partial differential equations to more complex objects and processes

- Model parameters are directly derived from the geometry of internals

Results obtained:

- Pertraction (liquid membrane extraction)
- Microdistillation
- Distillation in structured packed columns
- Reactive stripping in packed columns and monoliths

Fluid-dynamics approach

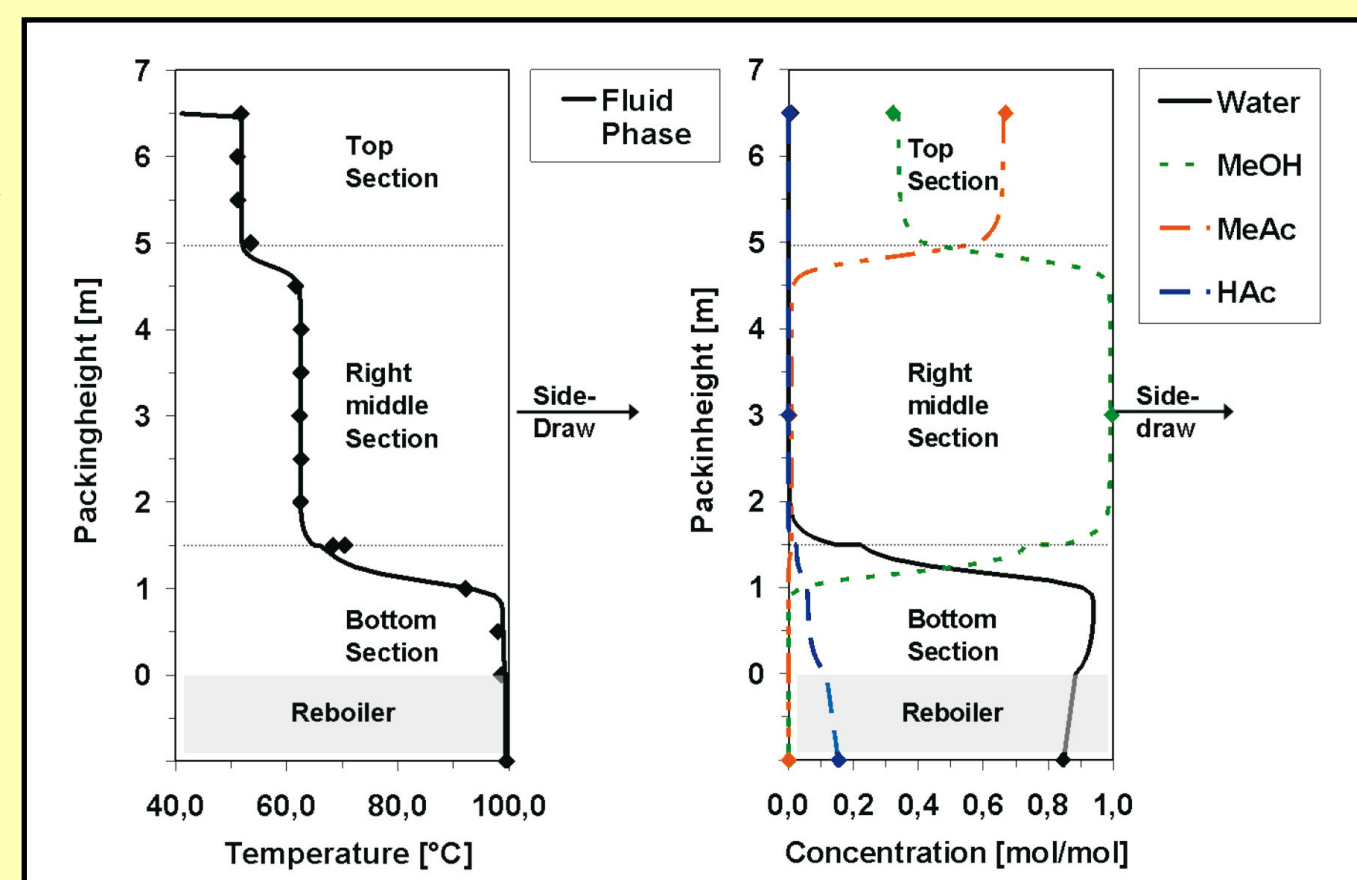
Idea: Column sub-division onto segments (stages) and their kinetics-based description

Characteristics:

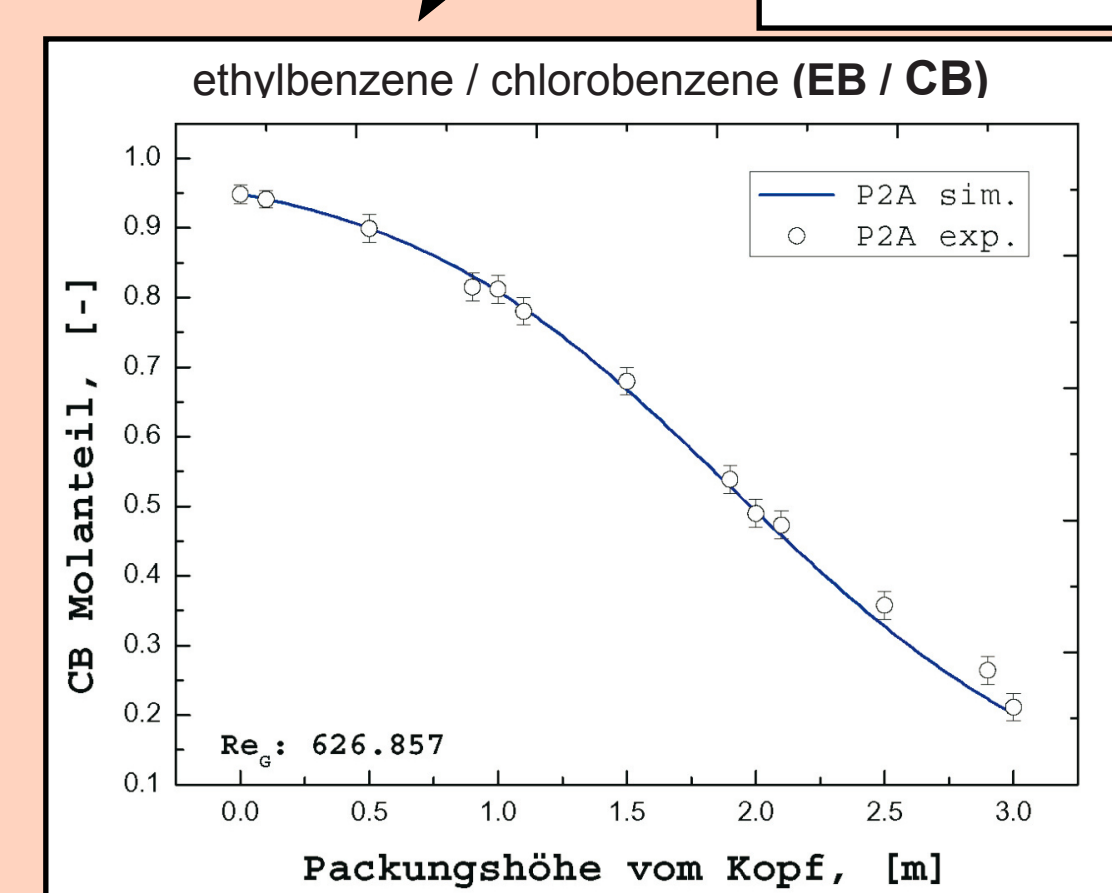
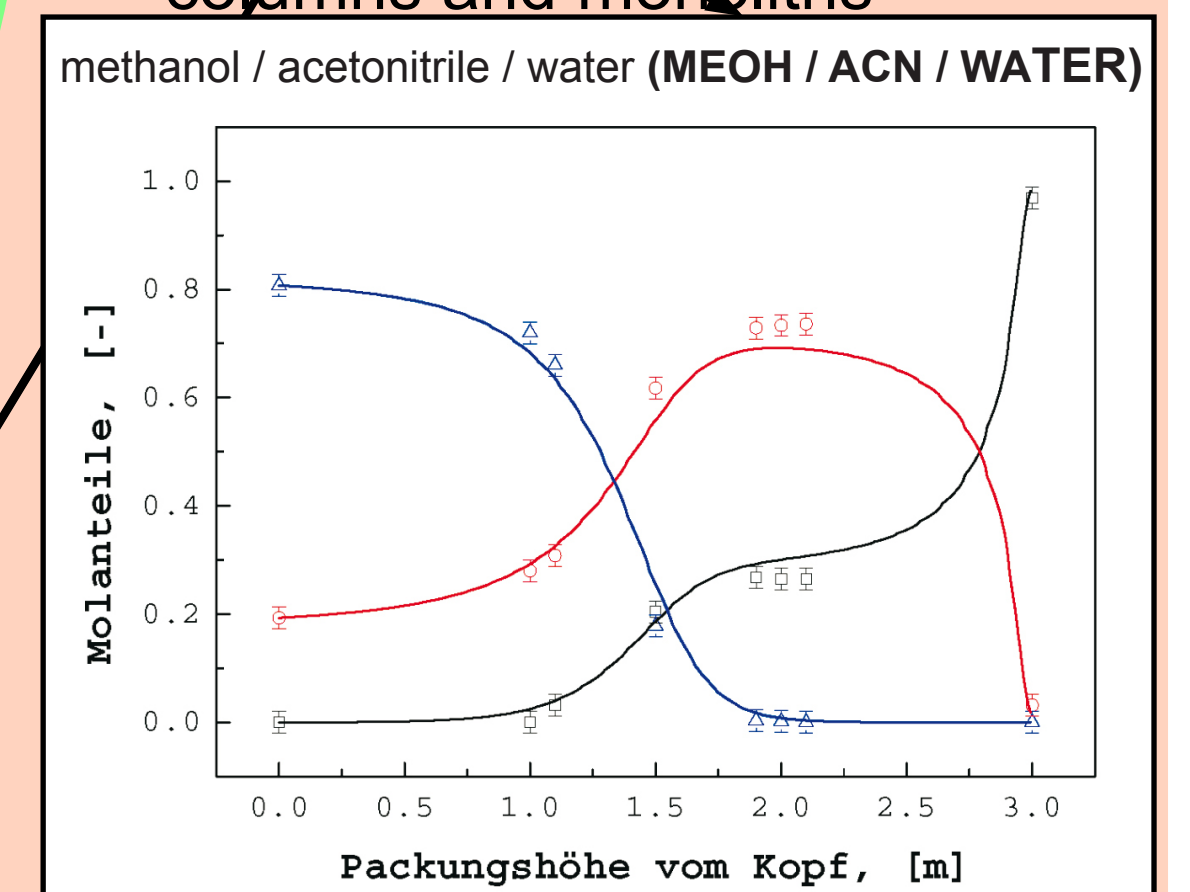
- Suitable in design and optimisation tasks for many staged operations, including reactive and hybrid separations
- Weak point is that accuracy and predictivity strongly depend on the quality of model parameters

Results obtained:

- Distillation and reactive distillation
- (Reactive) absorption and desorption
- Reactive Stripping
- Non-Reactive and reactive dividing wall columns
- Different Systems, units, internals
- Steady-state and dynamic simulations
- Optimisation studies



C. Grossmann, E. Y. Kenig, 2007, CIT plus, No. 5, 38-41



A. Shilkin & E. Y. Kenig 2005, Chem Eng J., 110,87-100
A. Shilkin, E. Y. Kenig, Z. Olujić, 2006, AlChE Journal, 52,3055-3066

Extension

Idea: Reaction kinetics measurements for CO₂ absorption into aqueous blends of alkanolamines prepared from renewable resources

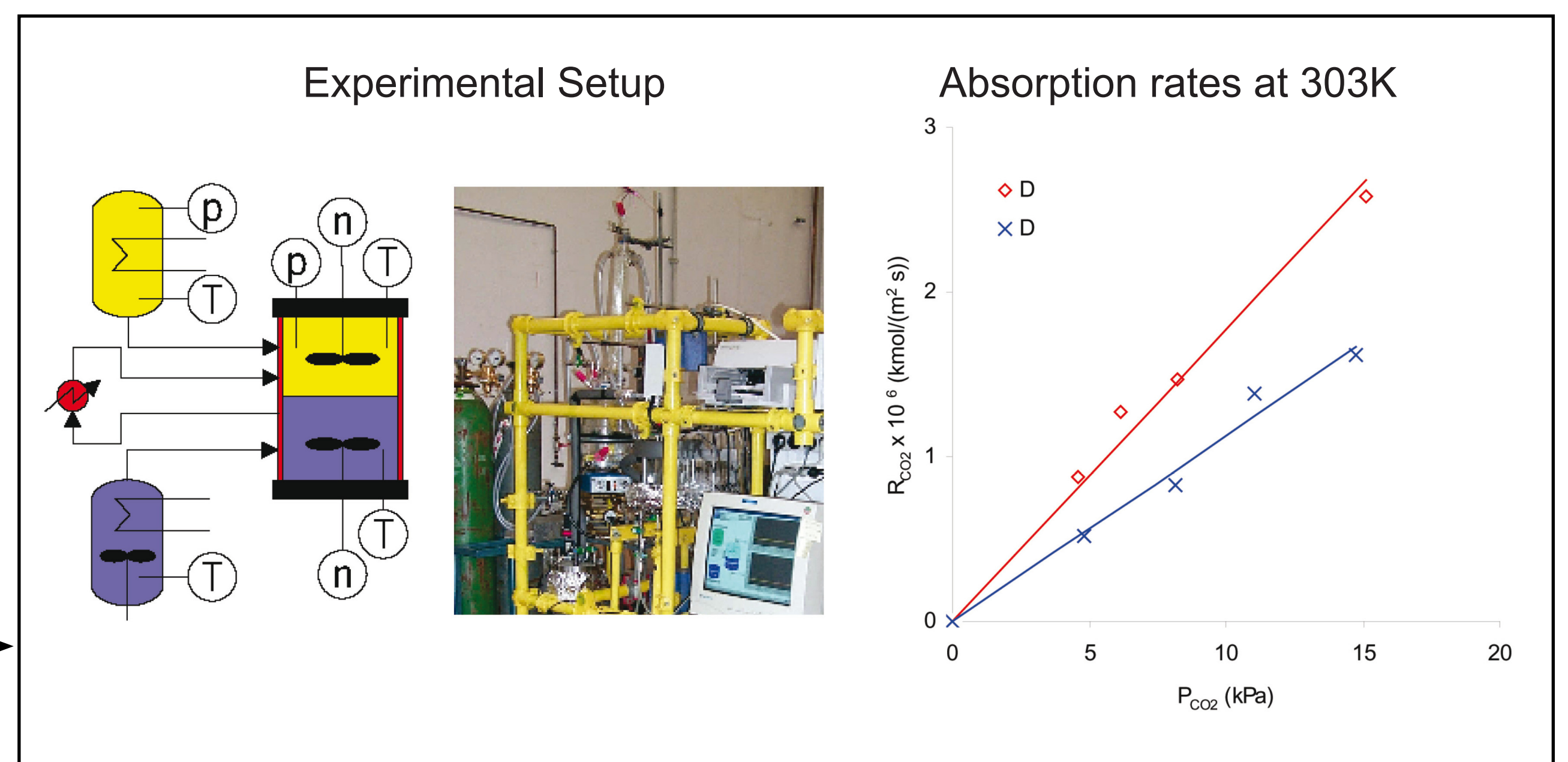
Characteristics:

- Data are obtained in a stirred cell reactor with a plane, horizontal gas-liquid interface
- Easy-to-use experimental device operated batch-wise
- The method is based on a simple fall-in-pressure technique,

without measurements of concentrations

Results obtained:

- Absorption of CO₂ into aqueous solutions containing N,N-diethylethanolamine (DEEA), N-ethylethanolamine (EEA) and their blends
- Reaction acceleration by piperazine (PIP)



P.D. Vaidya & E. Y. Kenig, 2007, Chem. Eng. Sci., 62, 7344-7350